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2. 3.40×10^{-3} _____ 5. 5.02×10^4 _____ 3. 2.67×10^5 _____ 6. 1.21×10^{-7} _____ Use your calculator to solve the following. Be sure to

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express your answer in scientific notation even if your calculator does not. 1. $(3.0 \times 10^3) + (5.0 \times 10^3)$ _____ 2. $(6.9 \times 10^3)(2.455 \times 10^4)$ _____ 3. $9^{-3} 5$

~~Worksheet #3: Scientific Notation Name~~

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What would be the mass of 5.3×10^{22} formula units of calcium iodide?

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9s Trends on the Periodic Table: Students learn to describe the pattern in atomic number, atomic mass, atomic radius, ionization energy, and electron affinity as they look across a period and down a family of the periodic table.

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The atomic nucleus is explored in this program. ...

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