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The de Broglie wavelength and speed of an electron. The De-

Broglie wavelength is $= 7.77 \times 10^{-11} \text{ m}$ The speed of an electron

is $= 9.376 \times 10^6 \text{ m} \cdot \text{s}^{-1}$ Given data: The electron accelerated with

a potential difference $(V) = 250 \text{ V}$ Formula used: $KE = \frac{1}{2} m v^2$

$= e V$ Where, $KE =$ kinetic energy $m =$ mass $e =$ charge of electron v

$=$ velocity $V =$ potential difference Re-arrange the equation, $v = \left(\frac{2 e V}{m} \right)^{1/2}$ is the required ...

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