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~~Chapter 8: Covalent Bonding – FCPS~~

Chapter 8: Covalent Bonding, a bond angle of 109.5 degrees that results when a central atom forms four bonds directed toward the center of a regular tetrahedron. the mixing of several atomic orbitals to form the same total number of equivalent hybrid orbitals.

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Answers Chapter 8 Covalent Bonding, chemical bond. covalent bond. molecule. electron dot diagram. |||the force that holds two atoms together. |||the chemical bond that results from sharing valence electro|. |||a molecule is formed when two or more atoms bond covalently. |||used to show valence electrons of

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Chapter 8 - Covalent Bonding - 8.4 Polar Bonds and Molecules - 8.4 Lesson Check - Page 253: 37 Answer If polar molecules are placed between oppositely charged plates, they tend to become attracted with respect to positive and negative plates.

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Q. Energy required to separate 1 molecule of ions from an ionic compound. The smaller the ions, the greater this energy is: Q.

~~Chapter 8/9: Ionic, Metallic, and Covalent Bonding (Final ...~~

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