

Chapter 17 The Atomic Nature Of Matter Exercise Answers

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CHAPTER 17 THE ATOMIC NATURE OF MATTER 327 17.2 Atoms Are Small Atoms are so small that there are about 1023 atoms in a gram of water (a thimbleful). The number 1023 is an enormous number, more than the number of drops of water in all the lakes and rivers of the world. So there are more atoms in a thimbleful of water than

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~~elix--CLix~~

Atoms are imperceptibly small, irreducible particles that make up all the ordinary matter of daily life. Most are older than the Earth and will live forever.

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88 Chapter 17 The Atomic Nature of Matter © Pearson Education, Inc., or its af? liate(s). All rights reserved. Subatomic Particles Three fundamental particles of the atom are the , , and . At the center of each atom lies the atomic which consists of and . The atomic number refers to the number of in the nucleus. All atoms of the same element

~~Concept-Development-17-4-Practice-Page~~

ATOMIC NATURE OF MATTER DOE-HDBK-1019/1-93 Atomic and Nuclear Physics Atomic Nature of Matter Summary Atoms consist of three basic subatomic particles. These particles are the proton, the neutron, and the electron. Protons are particles that have a positive charge, have about the same mass as a hydrogen atom, and exist in the nucleus of an atom. ...

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